Docket No.: 83068-US1

Claims 1 - 10 (previously cancelled)

Claim 11 (previously amended) A process for preparing a defective metal oxide for a battery

cathode with increased lithium capacity, said process comprising:

providing a sufficient amount of metal oxide;

heating said metal oxide under an atmosphere consisting essentially of O<sub>2</sub> and H<sub>2</sub>O<sub>(g)</sub> gas;

and

cooling said metal oxide, wherein said heating under said atmosphere introduces local ionic

defects and increases the lithium capacity of said metal oxide.

Claims 12-16 (previously canceled)

Claim 17 (previously amended): The process as in claim 11, further comprising the step of:

maintaining said heating at a temperature of from about 300 to about 600 °C.

Claim 18 (previously amended): The process as in claim 17, wherein said heating is maintained

from about 6 to about 72 hours.

Claim 19 (previously amended): The process as in claim 11, wherein said  $O_2$  and  $H_2O_{(g)}$  is applied

to said metal oxide sample at a linear flow rate of about 50 ccm to about 350 ccm.

2

Docket No.: 83068-US1

about 20 °C/min up to about 460 °C.

Claim 21 (previously presented): The process as in claim 20, further comprising the step of:

maintaining said temperature of about 460 °C for 24 hours.

Claim 22 (previously presented): The process as in claim 11, wherein said cooling is from about 2

to about 20 °C/min until ambient air temperature is achieved.

Claim 23 (previously amended): The process as in claim 11, wherein said metal oxide is V<sub>2</sub>O<sub>5</sub>.

Claim 24 (previously amended): The process as in claim 11, wherein said metal oxide comprises a

surface area of about 1-10 square meters.

Claim 25 (previously amended) A process for preparing a defective metal oxide for a battery

cathode with increased lithium capacity, said process comprising:

providing a sufficient amount of metal oxide;

heating said metal oxide at a temperature of from about 300 to about 600 °C for a time

period of from about 6 to about 72 hours under an atmosphere consisting essentially of O<sub>2</sub> and

 $H_2O_{(g)}$  gas, wherein said  $O_2$  and  $H_2O_{(g)}$  is applied to said metal oxide sample at a linear flow rate of

about 50 ccm to about 350 ccm; and

3

Application No. 10/046,295 Amendment dated November 26, 2008

Reply to Office Action of May 28, 2008

cooling said metal oxide, wherein said heating under said atmosphere introduces local ionic defects and increases the lithium capacity of said metal oxide.

Docket No.: 83068-US1